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Physical chemistry is the study of macroscopic, and particulate phenomena in chemical systems in terms of the principles, practices, and concepts of physics such as motion, energy, force, time, thermodynamics, quantum chemistry, statistical mechanics, analytical dynamics and chemical equilibrium.. Physical chemistry, in contrast to chemical physics, is predominantly (but not always) a ...

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Physical chemistry, Branch of chemistry concerned with interactions and transformations of materials. Unlike other branches, it deals with the principles of physics underlying all chemical interactions (e.g., gas laws), seeking to measure, correlate, and explain the quantitative aspects of reactions. Quantum mechanics has clarified much for physical chemistry by modeling the smallest particles ...

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Green chemistry principles 3, 4, 5, and 12 guide designers to reduce the hazards of chemicals. Principle 10, however, guides the design of products that degrade after their commercial function in order to reduce risk or the probability of harm occurring. Risk is a function of both a molecule's inherent hazard AND exposure – contact between a chemical and a species. Degradation can ...

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This course provides an introduction to the chemistry of biological, inorganic, and organic molecules. The emphasis is on basic principles of atomic and molecular electronic structure, thermodynamics, acid-base and redox equilibria, chemical kinetics, and catalysis. One year of high school chemistry is the expected background for this freshman-level course.

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[Principles of physical science | Britannica](#)

The green chemistry concept applies innovative scientific solutions to solve environmental issues posed in the laboratory. Paul T. Anastas, an organic chemist working in the Office of Pollution Prevention and Toxins at the EPA, and John C. Warner developed the Twelve Principles of Green Chemistry in 1991. These principles can be grouped into ...

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Physical Chemistry. Physical Chemistry is the branch of chemistry which is quite like mathematics and physics in developing underlying principles of chemistry. It observes and explains the physical and chemical properties of matter. Physical chemistry at IIT JEE level starts from basics of mole concept, quantities at atomic level and covers energy, thermodynamics and kinetics which forms the ...

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[Green chemistry - Wikipedia](#)

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[The Journal of Physical Chemistry C](#)

14 Organic Chemistry. A simple introduction to organic chemistry: nomenclature, functional groups, and some reaction types. 15 Inorganic Chemistry. An overview of some inorganic chemistry including complexation. (Physical States) Chapters 2 and 5 combined into one "chapter". Intermolecular forces govern the physical properties of matter. Gases ...

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and reorganized into discrete Topics, to make the text more flexible for teachers and more readable for students.

[The Journal of Physical Chemistry A](#)

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The change of state is likewise a physical change. In this scenario, one can observe a number of physical properties changing, such as viscosity and shape. As ice turns into water, it does not retain a solid shape and now becomes a viscous fluid. The physical

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"reaction" for the change of ice into liquid water is: $\text{H}_2\text{O}_{(s)} \rightarrow \text{H}_2\text{O} \dots$

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principles of vibrational spectroscopy and the origins of group frequencies. The book includes sections on basic principles in Chapters 1 and 2; instrumentation, sampling methods, and quantitative analysis in Chapter 3; a discussion of important environmental effects in Chapter 4; and a discussion of the origin of group frequencies in Chapter 5. Chapters 4 and 5 provide the essential ...

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